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Oxidizing and Reducing Agents. An oxidizing agent, or oxidant, gains electrons and is reduced in a chemical reaction. Also known as the electron acceptor, the oxidizing agent is normally in one of its higher possible oxidation states because it will gain electrons and be reduced.

Oxidizing and Reducing Agents - Chemistry LibreTexts

Positron Emission. Nuclides that are imbalanced in their ratio of protons to neutrons undergo decay to correct the imbalance. Nuclei that are rich in protons relative to their number of neutrons can decay by conversion of a proton to a neutron, emitting a positron (${}^0_1e^+$) and a neutrino (ν). Positrons are the antiparticles of electrons, therefore a positron has the same mass as an ...

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Acidic oxide - Wikipedia

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An intramolecular force (or primary forces) is any force that binds together the atoms making up a molecule or compound, not to be confused with intermolecular forces, which are the forces present between molecules. The subtle difference in the name comes from the Latin roots of English with inter meaning between or among and intra meaning inside. ...

Intramolecular force - Wikipedia

General Chemistry: Principles and Modern Applications (11th edition). Toronto: Pearson Canada Inc.) Seiring air es dipanaskan di atas titik lebur, pemutusan ikatan hidrogen terus berlanjut sehingga molekul-molekul air menjadi semakin tersusun rapat dan densitas air semakin meningkat.

Ikatan Hidrogen - Pembahasan Lengkap, Contoh, & Contoh Soal

2 David W. Oxtoby and Norman H. Nachtrieb Principles of modern chemistry, 2nd Ed. Saunders College Pub., 1990

Zhihu

Come in qualsiasi sistema tampone, il pH è bilanciato dalla presenza sia di un acido debole (ad esempio, H₂CO₃) che della sua base coniugata (ad esempio, HCO₃⁻), in modo da neutralizzare l'eventuale eccesso di acido o base introdotto nel sistema. Il mancato funzionamento di questo sistema provoca uno squilibrio acido-base, come acidemia (pH <7,35) e alcaliemia (pH > 7,45) nel sangue.

Sistema tampone bicarbonato - Wikipedia

^ 49.0 49.1 49.2 David W. Oxtoby; H. Pat Gillis; Alan Champion. Principles of Modern Chemistry, 7th ed.. Principles of Modern Chemistry, 7th ed.. Cengage Learning.

Etanol

(ethanol) (ethylalcohol), (C₂H₅OH, C₂H₆O, CH₃CH₂OH). È un liquido incolore, infiammabile, volatile, miscelabile con l'acqua.

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Hidrogenul este elementul chimic în tabelul periodic al elementelor cu simbolul H și numărul atomic 1. Este un gaz ușor inflamabil, incolor, insipid, inodor, iar în natură se întâlnește mai ales sub formă de moleculă diatomică, H₂. Având masa atomică egală cu 1,00794 u.a.m., hidrogenul este cel mai ușor element chimic. Etimologic, cuvântul hidrogen este o combinație a două ...

Hidrogen - Wikipedia

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História Origem e evolução. Segundo a teoria Alpher-Bethe-Gamov, o hidrogénio formou-se na época da nucleossíntese primordial, quando o Universo expandiu e arrefeceu o suficiente para permitir a ligação entre eletrões e protões. [13] O deutério formou-se por captura de neutrões. [13] A junção de mais neutrões e subsequente decaimento levou à formação de hélio e lítio.

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