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hydrogen atom is given by $\rho = q \delta^3(\mathbf{r} - \mathbf{r}_1) + \rho_2$ where q is the magnitude of the electronic charge, and $r_1 = a_0 = 2a_0$, a_0 being the Bohr radius. Find the distribution of charge (both continuous and discrete) that will give

Physics 505 Fall 2007 Homework Assignment #1 | Solutions

These are my solutions for problems from John David Jackson's Classical Electrodynamics (3rd Edition). Brace yourself — I did not get full marks on many of these. Getting any single Jackson problem completely correct could be a life's work!

Jackson: Electrodynamics | Ben Levy

Step-by-step solution: Chapter: CH1 CH2 CH3 CH4 CH5 CH6 CH7
CH8 CH9 CH10 CH11 CH12 CH13 CH14 CH15 CH16 Problem: 1 2
3(a) 3(b) 5 6(a) 6(b) 9 10(a) 10(b) 10(c) 14 16 1P 2P 3P 4P 5P 6P
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